

Single crystal structure of HMEDA-(DCA)₂, energetic hypergolic ionic liquid salts

Sung June Cho^{1,*}, Hoi-Gu Jang¹, Jin Hee Park¹, Gyeong Cheol Jo¹, Eun Mee Goh²

¹Department of Chemical Engineering,
Chonnam National University,
77 Yongbong-ro, Buk-gu, Gwangju 500-757,
Republic of Korea

²Agency for Defense Development, Yuseong
P.O. box 35, Daejeon 305-600 Korea

*E-mail: sjcho@chonnam.ac.kr

Accurate altitude and attitude control of the satellite is critical technology. Chemical propulsion system using hydrazine as a monopropellant has been the state of art system.[1] However, anhydrous hydrazine (> 97%) is very toxic because of high vapor pressure. Therefore, large research effort has been directed to find non-toxic fuel for the same purpose, especially such as energetic *hypergolic* ionic liquid for non-catalytic chemical propulsion.

Previously, TMEDA was converted into *N*¹,*N*¹,*N*¹,*N*²,*N*²,*N*²-hexamethylethane-1,2-diaminium dicyanamide (HMEDA-(DCA)₂) through the scheme shown in Figure 1. HMEDA-(DCA)₂ has been prepared using the simple quaternization and subsequent ion exchange of dicyanamide. The HMEDA-(DCA)₂ solution, 60 wt% had ID in the range of 100 msec with white fuming nitric acid as the oxidizer. Further, the ID of the HMEDA-(DCA)₂ solution was decreased to 20-30 msec upon introducing co-catalyst.[2]

In this work, the crystal structure of HMEDA-(DCA)₂ has been studied for better understanding of the decomposition mechanism. Single crystal data of HMEDA-(DCA)₂ was collected using synchrotron radiation. The charge flipping method was applied to obtain the structural solution using the *superflip* program implemented in JANA2006.

The structure of HMEDA-(DCA)₂ was found to be triclinic, *P*-1 with lattice parameters of *a*=6.427 Å, *b*=8.726 Å, *c*=14.069 Å, α =88.658°, β =87.874° and γ =71.712° in which the chemical composition was C₈H₂₂N₂-

2(C₂N₃) with *R*₁/*wR*₂ = 0.0867/0.2157 over 1472 reflections for *I* > 3σ(*I*). The largest difference peak was 1.06 and the deepest hole was -0.50 eÅ³. The goodness-of-fit index was 2.11. The final maximum (change/s.u.) was 0.0437.

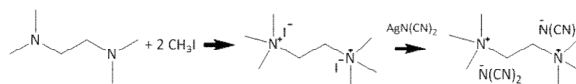


Fig.1 Schematic diagram for the synthesis of HMEDA-(DCA)₂ from TMEDA.

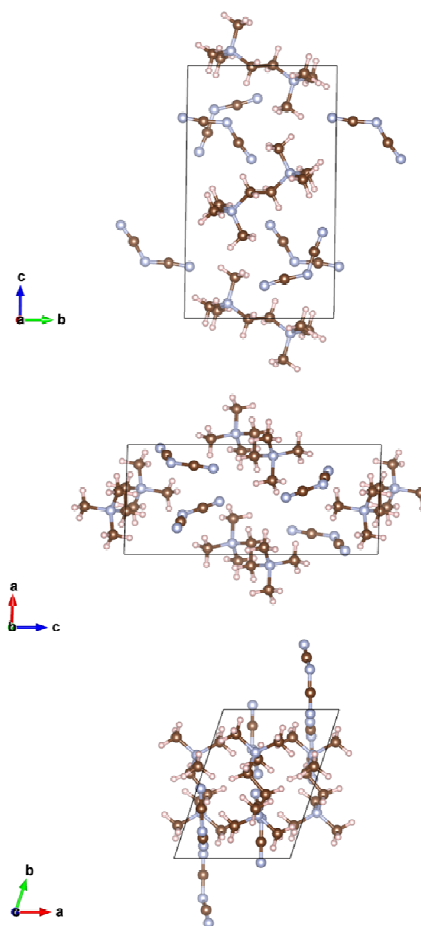


Fig.2 Crystal structure of HMEDA-(DCA)₂.

This research was supported by Defense Acquisition Program Administration and Agency for Defense Development (ADD 13010413).

REFERENCES

- [1] E.J. Wucherer, T. Cook, M. Stiefel, R. Humphries and J. Parker, 39th AIAA/ASME/SAE/ASEE Joint Propulsion Conference and Exhibit:5039 (2003).
- [2] G.C. Jo, E.M. Goh, S.W. Baek and S.J. Cho, Res. Chem. Intermed., 42 (2016) 201.