Co-catalysis of ReO$_x$-Au/CeO$_2$ and ReO$_x$/C in Hydrogenolysis of 1,4-Anhydroerythritol to 1,4-Butanediol

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Renewable biomass-based products are in increasing demand. However, as a chemical source, biomass contains higher oxygen content than fossil feedstocks. To deal with this drawback of biomass, the deoxydehydration (DODH) reaction would be a potential method to reduce the oxygen content in biomass products by removing vicinal diols in sugars simultaneously [1].

1,4-Anhydroerythritol can be produced by dehydration of erythritol which obtained from the fermentation of sugars. 1,4-Anhydroerythritol was able to be deoxygenated to 2,5-dihydrofuran and tetrahydrofuran in the DODH reaction over ceria-supported rhenium catalysts with gold and palladium promoters, respectively [2, 3]. In this study, we found that the physical mixture of ReO$_x$-Au/CeO$_2$ (Re = 1 wt %, Re/Au = 3 molar ratio) and carbon-supported rhenium (Re = 3%) catalysts showed high activity and good selectivity of 1,4-butanediol in the ring-opening reaction of 1,4-anhydroerythritol.

Various supported Re catalysts were tested for the co-catalysis with ReO$_x$-Au/CeO$_2$. Among them, the carbon black Black Pearls 2000 (BP2000), denoted as “C”, was the best support to obtain 1,4-butanediol. Figure 1 showed the time course of ring-opening reaction of 1,4-anhydroerythritol over the mixture of ReO$_x$-Au/CeO$_2$ and ReO$_x$/C. 1,4-Anhydroerythritol = 0.5 g, 1,4-butanediol = 8.0 MPa, $T$ = 413 K. A: tentatively assigned as (Fig. 1) 

![Fig. 1 Time course of the ring-opening reaction of 1,4-anhydroerythritol over the mixture of ReO$_x$-Au/CeO$_2$ and ReO$_x$/C. 1,4-Anhydroerythritol = 0.5 g, 1,4-butanediol = 8.0 MPa, $T$ = 413 K. A: tentatively assigned as.](Image)

Scheme. 1 Proposed reaction route from 1,4-anhydroerythritol to 1,4-butanediol over the mixture of ReO$_x$-Au/CeO$_2$ and ReO$_x$/C.

REFERENCES