

Photoelectrochemical Water Splitting with SrNbO₂N Photoanode Prepared by a Novel Roll-press Method

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Hydrogen produced via a photocatalytic and a photoelectrochemical route has been paid large attention to as a potential clean fuel. SrNbO₂N is an attractive materials due to its long absorption edge of 690 nm which covers a wide range of visible light region[1]. In this study, a novel electrode preparation method, a roll-press method, was applied to SrNbO₂N photoanodes. The effects of preparation conditions and post treatments were investigated.

SrNbO₂N particles were synthesized by nitridation of Sr₂Nb₂O₇ oxide precursors under an NH₃ flow[2]. Obtained particles were drop-casted on a Nb substrate followed by roll-pressing as shown in Fig. 1 (Left). After pressing, pressed films were post-annealed under a N₂ flow at different temperature for 30 min. Photoelectrochemical properties were investigated in a typical three-electrode configuration.

After roll-pressing, particles were crushed and grain boundaries became unclear as shown in Fig. 1 (Right). Figure 2 shows current-potential curves of SrNbO₂N/Nb prepared by a roll-press method modified by CoPi as a cocatalyst[3]. Without post heating under N₂ flow, only several micro ampere of photocurrent was observed. According to increasing the annealing temperature, clear photocurrents appeared and photocurrent over 0.5 mA/cm² at 1.2 V_{RHE} was observed when heated at 873 K. It is more likely that annealing reduced the electrical resistance between particles and substrates. On the other hand, annealing at higher temperature reduced photocurrent probably due to the degradation of SrNbO₂N particles at a high temperature.

As a brief summary, SrNbO₂N/Nb exhibited clear photoanodic current, indicating that the roll-press method is a promising way for electrode preparation. In the presentation, effects of other pre- and post-treatment on SrNbO₂N/Nb will be discussed.

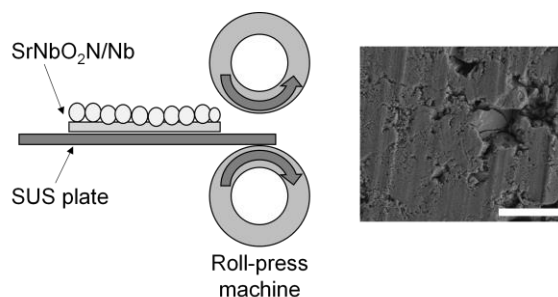


Fig.1 (Left) Schematic of a roll-press method. (Right) Top-view SEM image of SrNbO₂N/Nb electrode. Scale bar indicates 5 μm.

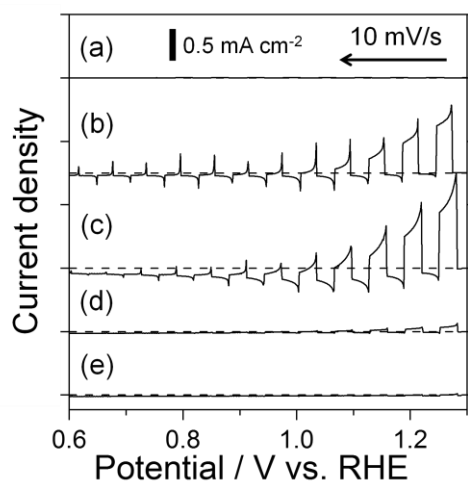


Fig. 2 Current-potential curves of SrNbO₂N/Nb photoelectrodes prepared by a roll-press method (a) w/o post-heating and (b)-(e) with different post heating temperatures. (b) 773 K, (c) 873 K, (d) 973 and (e) 1073 K. AM 1.5G, 1 M KBi solution (pH 10).

REFERENCES

- [1] H. Urabe, T. Hisatomi, T. Minegishi, J. Kubota, and K. Domen, *Faraday Discuss.*, 176 (2014) 213–223.
- [2] M. Kodaera, H. Urabe, M. Katayama, T. Hisatomi, T. Minegishi, and K. Domen, *J. Mater. Chem. A*, 4 (2016) 7658–7664.
- [3] M. W. Kanan and D. G. Nocera, *Science*, 321 (2008) 1072–1075.