## Study on the effect of natural minerals on the product gas composition from gasification of lignite under co-feeding of steam and CO<sub>2</sub>

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Coal Gasification has been investigated all over the world to develop innovative and ecofriendly technologies that could reduce the emission of environmental pollutant. Most attention on coal gasification have been paid to with either steam or  $CO_2$  atmosphere [1]. And, many gasification studies are performed in a fluidized bed reactor, where bed materials are mixed with the catalyst to improve the reactivity of the fluidized bed reactor [2].

In this study, the effect of natural minerals addition was investigated when gasification was carried out with steam and  $CO_2$  mixture.

The dolomite and kaolin were used as natural minerals. 5 wt % of natural mineral was physically mixed with lignite. Steam and CO<sub>2</sub> mixed gasification reaction system was carried out with micro quartz reactor at various temperatures of 900 °C. The product gases were analyzed continuously with gas analyzer and gas chromatography.

Table 1 shows the total moles of  $H_2$ , CO, and  $CO_2$  produced and the carbon conversion at 900 °C.

As the composition of  $CO_2$  in the reaction gas increased, the amount of product gas  $CO_2$  decreased and the value of  $(H_2 + CO) / CO_2$  increased.

When comparing the values of  $(H_2 + CO) / CO_2$  of the syngas produced in the same reaction gas composition, the highest except for steam 300 CC/min were shown with Dolomite 5 wt%.

Fig. 1 shows the gas production profile over time according to each reaction gas conditions at 900 °C. The time to reach the carbon conversion of 80 % was 8 to 14 minutes. Especially, when  $CO_2$  100 cc/min + steam 200 cc/min was used as the reaction gas, the gasification reaction was terminated at the fastest.

As a result, the amount of  $H_2$  increased when the composition of steam was increased in the reaction gas, and the fastest gasification reaction was observed when  $CO_2$  100 cc/min + steam 200 cc/min was used as the reaction gas.

	H <sub>2</sub> (mmol)	CO (mmol)	CO <sub>2</sub> (mmol)	H2+CO CO2	Carbon conversion (%)	H <sub>2</sub> (mmol)	CO (mmol)	CO <sub>2</sub> (mmol)	H2+CO CO2	Carbon conversion (%)
NA Catalysty	Steam 300 cc/min					CO2 100 cc/min + Steam 200 cc/min				
	13.51	4.48	11.41	1.58	89.5	10.2	6.03	10.53	1.54	92.55
	CO2 200 cc/min + Steam 100 cc/min					CO <sub>2</sub> 300 cc/min				
	5.84	10.45	5.79	2.81	90.81	0	13.06	2.76	4.73	89.59
Dolominte 5 wt%	Steam 300 cc/min					CO2 100 cc/min + Steam 200 cc/min				
	14.13	4.75	12.53	1.51	93.15	11.1	6.42	10.85	1.61	95.54
	CO2 200 cc/min + Steam 100 cc/min					CO <sub>2</sub> 300 cc/min				
	6.11	10.79	5.83	2.90	92.69	0	13.57	2.86	4.74	92.05
Kaalin 5 wt%	Steam 300 cc/min					CO2 100 cc/min + Steam 200 cc/min				
	13.83	4.01	12.31	1.45	91.53	10.98	6.02	10.53	1.61	93.78
	CO2 200 cc/min + Steam 100 cc/min					CO <sub>2</sub> 300 cc/min				
	6.02	10.79	6.13	2.74	91.72	0	13.34	3.18	4.19	90.65

Table 1 Total yields of major product gases



Fig. 1 Graph of Time vs Gas production

## REFERENCES

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