## Direct carbamates synthesis from CO<sub>2</sub>, amines and alcohols using CeO<sub>2</sub> and 2-cyanopyridine

<u>Yu Gu<sup>1</sup></u>, Ayaka Miura<sup>1</sup>, Masazumi Tamura<sup>1</sup>, Yoshinao Nakagawa<sup>1</sup>, Keiichi Tomishige<sup>1</sup> <sup>1</sup>School of Engineering, Tohoku University, Sendai, Japan \*E-mail: tomi@erec.che.tohoku.ne.jp

Carbamates are compounds of great interest owing to their application in organic synthesis. Methyl-*N*-phenylcarbamate (MPC) is an important intermediate for the synthesis of polyurethane. various Among synthesis methods of MPC, direct synthesis of MPC from CO<sub>2</sub>, aniline and methanol will be an eco-friendly process because this process uses cheap and abundant CO<sub>2</sub> as a carbonyl source and H<sub>2</sub>O is the only byproduct. Generally, the MPC yield is very low in this reaction due to the severe equilibrium limitation and the low reactivity of aniline [1]. Recently, we found that  $CeO_2 + 2$ -cyanopyridine catalyst system is effective for direct synthesis of organic carbonates from CO<sub>2</sub> and alcohols [2]. In this study, we applied this catalyst system to the direct synthesis of carbamates.

MPC synthesis from  $CO_2$ , aniline and investigated 2methanol was with cyanopyridine as a reaction promotor and various metal oxides catalysts, which showed that CeO<sub>2</sub> was the most effective metal oxide among the metal oxides tested. With combination of CeO<sub>2</sub> and 2-cyanopyridine, high MPC yield of 95% was achieved at 423 K for 6 h. In this reaction, dimethyl carbonate (DMC) and 2-picolinamide were coproduced as main byproducts by carboxylation of methanol with CO<sub>2</sub> and hydration of 2cyanopyridine, respectively (Scheme 1). The produced amount of 2-picolinamide was comparable to the produced amount of H<sub>2</sub>O by MPC and DMC formation. This result showed that 2-cyanopyridine efficiently worked as a dehydrating agent and shifted the chemical equilibrium to the carbonate and carbamate side.

Fig. 1 shows the time-course of the MPC synthesis over  $CeO_2$  with 2-cyanopyridine. High *N*,*N*`-diphenylurea (DPU) selectivity of 74% was detected at 0 h and the selectivity decreased with the reaction time. At the same time, MPC selectivity increased from 26% at 0 h to 97% at 8 h. This result suggested that MPC was produced via formation of DPU.

In addition, the combination of  $CeO_2$  and 2cyanopyridine was applied to reactions of various alcohols and amines affording the corresponding carbamates in high yields.



Scheme 1 MPC synthesis from  $CO_2$ , aniline and methanol by  $CeO_2 + 2$ -cyanopyridine system: (1) carboxylation of  $CO_2$ , aniline and methanol, (2) hydration of 2-cyanopyridine to 2-picolinamide, (3) carboxylation of methanol with  $CO_2$ .



Fig. 1 The direct synthesis of MPC from  $CO_2$ , aniline and methanol using  $CeO_2$  and 2-cyanopyridine. Reaction condition: aniline 5 mmol, methanol 75 mmol, 2-cyanopyridine 75 mmol,  $CO_2$  5 MPa,  $CeO_2$  1 mmol, reaction temperature 423 K.

## REFERENCES

[1] M. Honda, S. Sonehara, H. Yasuda, Y. Nakagawa and K. Tomishige, Green Chem., 13 (2011) 3406.

[2] M. Honda, M. Tamura, Y. Nakagawa, K. Nakao, K. Suzuki and K. Tomishige, J. Catal., 318 (2014) 95.