

Development of Novel Dehydrogenation Catalyst for Hydrogen Carrier System

Yoshimi Okada,^{a,*} Kenichi Imagawa,^b Makoto Yasui^c

^a*Technology Development Department, 4-6-2 minatomirai, Nishi-ku, Yokohama, 220-8765, Japan*

^b*R&D Center, 3-13 Moriya-cho, Kanagawa-ku, Yokohama, 221-0022, Japan*

^c*Global Environment & Green Energy Project Operations, 4-6-2 minatomirai, Nishi-ku, Yokohama, 220-8765, Japan*

**Corresponding author: +81-45-225-4986, okada.yoshimi@chiyodacorp.com*

Abstract: Chiyoda Corporation developed “SPERA HydrogenTM” system for the massive H₂ storage and transportation technology through a pilot plant demonstration. The system employs the Organic Chemical Hydride method (OCH method). In the method, hydrogen is fixed to toluene and converted to methylcyclohexane (MCH) as a Liquid Organic Hydrogen Carrier (LOHC). Since hydrogen is stored in the MCH molecule, hydrogen is able to be stored and transported under the ambient temperature and pressure in the liquid state. Hydrogen gas volume is reduced around bellow than 1/500. If we reduce the volume physically, we need more than 50MPa, it is realized under the ambient condition by using chemical reaction in this method. The key technology of “SPERA HydrogenTM” system was a development of a novel dehydrogenation catalyst which is needed for hydrogen generation after transportation at the hydrogen utilization place, since there was not catalyst with stable performance for MCH dehydrogenation due to severe coking for the catalyst deactivation. The developed novel dehydrogenation catalyst is nano-platinum particle on the alumina carrier in a uniform type. In this paper, we introduce the development of the dehydrogenation catalyst, outline of “SPERA HydrogenTM” system and future prospect for hydrogen utilization in each sector including CCU (Carbon Dioxide Capture and utilization).

Keywords: nano-platinum catalyst, Dehydrogenation, Hydrogen, Storage, Transportation.

1. Introduction

We are facing to the climate change issue by carbon emission, we have to promote a decarbonization of energy system. Hydrogen energy has much potential for a solution of the issue. Japanese government issued Strategic Road Map of Hydrogen and Fuel Call in 2014¹⁾, and it was revised in 2016. In addition, Basic Hydrogen Strategy also issued in last December²⁾. The 5th Basic Energy Plan of Japan will be issued in this summer³⁾, the Hydrogen energy will be mentioned in it based on the Basic Hydrogen Strategy. The Ministerial Meeting on Hydrogen Energy will be held in Japan in this year on 23rd Oct. ⁴⁾

Since the massive hydrogen storage and transportation technology is needed for hydrogen energy system, brisk R&D for hydrogen energy carrier is promoted in these years. Chiyoda Corporation has been completed a technical development of “SPERA HydrogenTM” system for the massive H₂ storage and transportation technology through a pilot plant demonstration. The system employs the Organic Chemical Hydride method (OCH method). In the method, hydrogen is fixed to toluene and converted to methylcyclohexane (MCH) as a Liquid Organic Hydrogen Carrier (LOHC). Toluene and MCH are gasoline components and in the liquid phase under the ambient temperature and pressure. In the method, the potential risk for the massive hydrogen storage and transportation can be reduced to the conventional one for the gasoline storage and transportation, since hydrogen is stored as MCH in the liquid phase under the ambient condition as same as gasoline. It is considered that the conventional storage tanks and the chemical tankers which are conventional infrastructure can be used to this system is also the merit. We named the developed system “SPERA HydrogenTM” system. The key technology of “SPERA HydrogenTM” system was a novel dehydrogenation catalyst development. The dehydrogenation catalyst is nano-platinum particle on alumina carrier in a uniform type. In this paper, we introduce the development of the dehydrogenation catalyst and “SPERA HydrogenTM” system.

2. SPERA Hydrogen System

2-1. Dehydrogenation Catalyst

We employ Toluene/Methylcyclohexane (MCH) system as the LOHC. Equation (1) shows the hydrogenation (H_2 storage reaction) and dehydrogenation (H_2 generation reaction).

The method had been investigated in the Euro-Quebec project as the third candidate with liquefied hydrogen and liquefied ammonia method in 1980's. In those days, the dehydrogenation catalyst life was only 1 or 2 days due to severe coking. Chiyoda started the catalyst development since 2002, and completed the catalyst development including industrial production of the catalyst in 2011.

Fig.1 shows the estimated surface model of the developed dehydrogenation catalyst⁵⁾. The catalyst is partially sulfided nano-sized Pt cluster on the Al_2O_3 catalyst. The catalyst has very high performance more than 10,000hrs catalyst life with more than 95% of MCH conversion and more than 99% of toluene selectivity in the MCH dehydrogenation reaction in the laboratory test. The nano-sized Pt cluster has dramatically improved the activity and catalytic life. The role of sulfur is considered to prevent the decomposition of MCH on the Pt cluster. Fig.2 shows the results of the measurement for the dispersion of sulfur and platinum atoms in the catalyst with EPMA (Electron Probe Micro Analyzer). The developed catalyst has a uniform dispersion of sulfur and Platinum atoms.

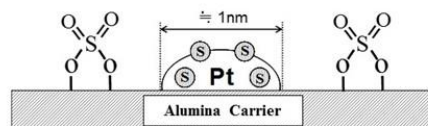
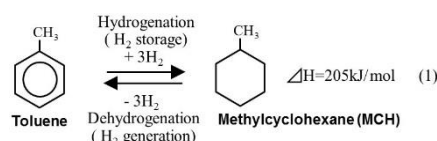


Fig.1 Estimated surface model of dehydrogenation catalyst

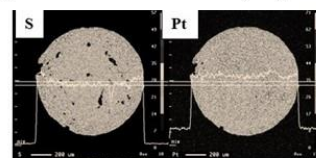


Fig.2 Result of S and Pt dispersion measurement by EPMA

2-2. Technical demonstration

Fig.3 shows the pilot plant for the technical demonstration with $50\text{Nm}^3\text{-H}_2/\text{h}$ capacity⁶⁾. In the demonstration, both of hydrogenation and dehydrogenation were operated at the same location. The operation was started in Apr. 2013, and completed in Nov. 2014 with highly stable performance for around totally 10,000hrs. Since the heat exchanger type reactor are employed for the both reaction, the scale up is relatively easy.

Fig.4 shows the reaction results of hydrogenation and dehydrogenation. In the initial period, it was operated at the several conditions to get the parameters for a development of the simulator. The reaction pressure is lower than 1MPa in the both reaction.

The hydrogenation temperature is lower than 250°C and the dehydrogenation temperature is lower than 400°C . In the Fig.4, the both conversions are shown in the ratio to the equilibrium conversion. The results of the both reaction shows the well stable performance. The hydrogenation process performance is toluene conversion: $>99\%$, MCH selectivity: $>99\%$, MCH yield: $>99\%$. The dehydrogenation process performance is MCH conversion: $>95\%$, toluene selectivity: $>99\%$, hydrogen and toluene yield: $>95\%$. If the dehydrogenation conversion declined to 95%, catalyst will be exchanged and Pt on the catalyst will be recovered from the spent catalyst for the fresh catalyst preparation^{7,8)}.

Now, we are executing the first international hydrogen supply chain demonstration project for hydrogen transportation from Brunei Darussalam in South East Asia to Kawasaki City in Japan in 2020 as a member of AHEAD (Advanced Hydrogen Energy Chain Association for Technology Development)⁹⁾. AHEAD is organized by Chiyoda Corporation, Mitsubishi Corporation, Mitsui Co& Ltd. and Nippon Yusen (NYK) and it is founded by NEDO (New Energy and Industrial Technology Development Organization) which is governmental organization.



(a) Reaction section



(b) Tank yard

Figure 3 Pilot plant for "SPERA Hydrogen" System

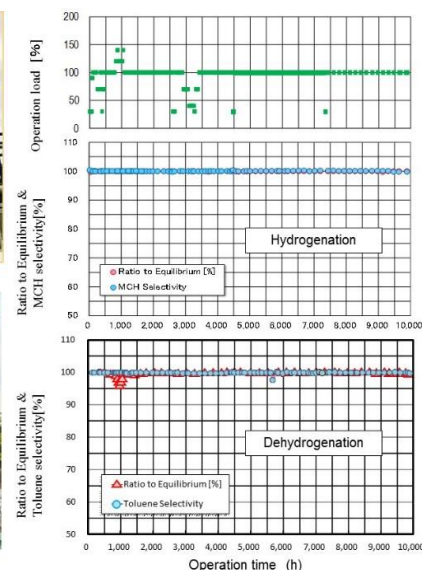


Figure 4 Results of demonstration operation

3. Prospects for Hydrogen Utilization

Figure 5 shows the future prospects for hydrogen utilization in each sector. Fuel cell vehicle (FCV) and residential fuel cell are already commercialized. It is highly effective to reduce CO₂ emission from car or family house which are small and decentralized CO₂ emission resources. It is necessary to reduce the CO₂ emission from large emission sources such as power generation, City gas and industrial sectors.

Since the power generation sector is the biggest emission source, we need to utilize hydrogen as generation fuel. There two ways for it. One is hydrogen will be used for the thermal power generation, the other is fuel for fuel cells as the decentralized power generation. The several MW class fuel Cells are viable for the airport, station or city hall which should be worked under the blackout due to the earth quake etc. In the city gas sector, hydrogen will be mixed to natural gas as Hydrogen methane which is called Hytan in the first step, and final way is considered that hydrogen produced from renewable energy and recovered CO₂ from large emission source are react to CO and H₂O through the reverse shift reaction, and then CO and H₂ are reacted to CH₄ through the sabatier reaction. In the industrial sector, recovered CO₂ will be sequestered as CCS (Carbon dioxide Capture and Sequestration) or renewable hydrogen will be react recovered CO₂ as CCU (Carbon dioxide Capture and Utilization).

CCS and CCU will be significant way as physical and chemical carbon reduction in the future. In addition, NH₃ from renewable hydrogen will be viable for CCU as recovered CO₂ and NH₃ are react to (NH₂)₂CO which is fertilizer¹⁰. This concept means emitted CO₂ from industry is go back to the natural carbon cycle by using N₂ as intermediate. Since the present NH₃ is produced from natural gas, NH₃ from renewable energy is also significant for the food issue by increasing population and gas depletion in the future.

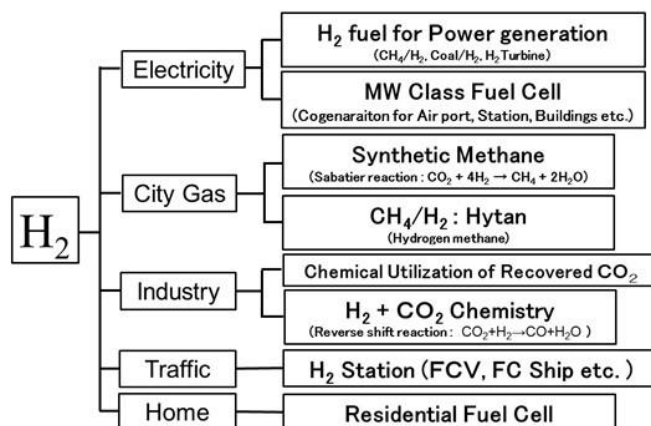


Figure 5 Prospects for Hydrogen Utilization

4. Acknowledgement

The first international hydrogen supply chain demonstration project by AHEAD is supported by NEDO. We express words of gratitude to NEDO.

5. Conclusion

Chiyoda completed the technical demonstration through the pilot plant demonstration. Now, we are executing the first international hydrogen supply chain demonstration project as a member of AHEAD. We would like to contribute to establishment of low carbon society, through the promotion of the hydrogen energy systems and reduction of CO₂ emission to prevent the global warming issue with the commercialization of “SPERA Hydrogen” system.

References

1. Strategic Road Map of Hydrogen and Fuel Cell, Ministry of Economy, Trade and Industry (METI), 2014 http://www.meti.go.jp/english/press/2014/0624_04.html
2. Basic Hydrogen Strategy, Ministry of Economy, Trade and Industry (METI), 2017 http://www.meti.go.jp/english/press/2017/1226_003.html
3. Basic Energy Plan of Japan, Ministry of Economy, Trade and Industry (METI), 2014 <http://www.enecho.meti.go.jp/en/>
4. Ministerial Meeting on Hydrogen Energy, Ministry of Economy, Trade and Industry (METI) <http://www.meti.go.jp/press/2018/05/20180508001/20180508001.html>
5. Y. Okada, K. Imagawa, M. Shimura, Fuel Cell Technology, **14** (2014) 36-40.
6. Y. Okada, Y. Hosono, J. of Japan Society for safety engineering, **53** (2014) 386-392.
7. Y. Okada, K. Imagawa, T. Mikuriya, M. Yasui, Shokubai, **57** (2015) 8-13.
8. Y. Okada, K. Imagawa, H. Kawai, PETROTECH, **38** (2015) 660-664
9. Chiyoda Corporation, www.chiyodacorp.com
10. Y. Okada, Energy and Resource, Vol.39, No.5 p.30 (2018).