# Morphologically uniform Co<sub>3</sub>O<sub>4</sub> hexagonal plates of the (112) facets with surface Fe/Mn doping and Au loading: Exceptionally active for catalytic combustion of benzene

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### 1. Introduction

Recent studies demonstrated that morphology or facet structure control is a powerful approach to achieving highly active, selective, yet durable catalyst systems for various applications. Catalytic removal of VOCs is a very important topic for environmental protection. Developing highly efficient Co-based catalyst becomes a hot spot for this kind of reaction. In the present study, we first synthesized the hexagonal plates of  $\beta$ -Co(OH)<sub>2</sub> precursor and further obtained the morphologically uniform Co<sub>3</sub>O<sub>4</sub> hexagonal plates of the (112) facets which is already very active for the target reaction. Simple doping of the structurally defined Co<sub>3</sub>O<sub>4</sub> substrate with Fe or Mn element can yield the morphology-retained Mn<sub>x</sub>Co<sub>3-x</sub>O<sub>4</sub> and Fe<sub>y</sub>Co<sub>3-y</sub>O<sub>4</sub> hexagonal plates with enhanced catalyst performance. With following precise Au deposition (~ 3 nm) of narrow particle size distribution, extremely active Au/Mn<sub>x</sub>Co<sub>3-x</sub>O<sub>4</sub> and Au/Fe<sub>y</sub>Co<sub>3-y</sub>O<sub>4</sub> are achievable with the lowest T<sub>100</sub> of 195 °C known to date for benzene combustion.

## 2. Experimental

(1) Synthesis of  $\beta$  -Co(OH)<sub>2</sub> precursor: CoCl<sub>2</sub>·6H<sub>2</sub>O of 7.12 g was dissolved in 100 ml distilled water under a N<sub>2</sub> flow (100 ml min<sup>-1</sup>). After that, NaOH of 4 g was dissolved in 40 ml water and the solution was added drop wise into the first solution under stirring. The mixture was divided into two parts and subjected to a hydrothermal treatment at 180 °C for 24 h. The precipitate was collected and washed with distilled water. (2) Synthesis of Co<sub>3</sub>O<sub>4</sub> hexagonal plates: Air calcination of  $\beta$ -Co(OH)<sub>2</sub> precursor was performed at 400 °C. (3) Synthesis of Mn<sub>x</sub>Co<sub>3-x</sub>O<sub>4</sub> and Fe<sub>y</sub>Co<sub>3-y</sub>O<sub>4</sub> hexagonal plates: In preparation of  $\beta$ -Co(OH)<sub>2</sub> precursor, part of CoCl<sub>2</sub> was replaced by MnCl<sub>2</sub> or FeCl<sub>2</sub>, with the molar ratio of Mn (or Fe) : Co = 3:97, 6:94, 9:91 and 12:88. The samples M<sub>3</sub>Co<sub>97</sub>O<sub>x</sub>, M<sub>6</sub>Co<sub>94</sub>O<sub>x</sub>, M<sub>9</sub>Co<sub>93</sub>O<sub>x</sub>, and M<sub>12</sub>Co<sub>88</sub>O<sub>x</sub> (M = Mn or Fe) were received. (4) Preparation of Au-loaded samples: Deposition-precipitation method was employed to load Au species on Co<sub>3</sub>O<sub>4</sub> plates and doped substrates with a nominal Au loading being 3% by weight. The product was collected by Centrifugation, dried at RT overnight, and calcined at 250 °C for 2 h. The samples were characterized by XRD, SEM, TEM, H<sub>2</sub>-TPR and evaluated for benzene combustion.

#### 3. Results and discussion

A series of  $Co_3O_4$ ,  $Mn_xCo_{3-x}O_4$ , and  $Fe_yCo_{3-y}O_4$ , together with the Au-deposited interfacial structures were achieved via the controllable synthesis approaches. The primarily exposed facet was identified to be (112) over pure  $Co_3O_4$  hexagonal plates, which exhibited excellent catalytic activity for benzene combustion ( $T_{100}$ = 240 °C). When the pure  $Co_3O_4$  hexagonal plates were doped with Mn or Fe element, even with a low content, the resulting  $M_3Co_{97}O_x$ ,  $M_6Co_{94}O_x$ ,  $M_9Co_{93}O_x$ , and  $M_{12}Co_{88}O_x$  (M = Mn or Fe) would not only maintain the original sample morphology but also improve catalyst performance. Among them,  $Mn_6Co_{94}O_x$ and  $Fe_6Co_{94}O_x$  were found to be the most effective for the reaction.  $T_{100}$  was notably reduced to 215 °C and 225 °C respectively for  $Mn_6Co_{94}O_x$  and  $Fe_6Co_{94}O_x$ . With precise Au loading over  $Fe_6Co_{94}O_x$ , the catalytic activity of benzene combustion was further considerably enhanced, with the  $T_{100}$  be reduced to 195 °C over  $Au/Fe_6Co_{94}O_x$ . TEM images shown in Fig. 1 demonstrated that although the Mn or Fe doping does not cause morphology variation of sample but significantly decreased the plate dimension, resulting in a remarkable increment in surface area of substrate. (HR)TEM image of the Au-loaded sample indicated that Au deposition had little impact on the facet structure of substrate. The doping element, however, showed direct influence on the interfacial catalysis of  $Au/M_6C_{094}O_x$  system.



Scheme 1. The preparation of Co<sub>3</sub>O<sub>4</sub> hexagonal plates and benzene combustion over Au/Co<sub>3</sub>O<sub>4</sub>.



Figure. 1. (A1-A3) SEM, TEM, HRTEM of Co<sub>3</sub>O<sub>4</sub> hexagonal plates, respectively. (A4-A5) HRTEM of Au/Co<sub>3</sub>O<sub>4</sub>, (B) ball-stick model (left) and polyhedral model (right) of cell structure of spinel, (C1-C4) SEM images of Mn<sub>3</sub>Co<sub>97</sub>O<sub>x</sub>, Mn<sub>6</sub>Co<sub>94</sub>O<sub>x</sub>, Mn<sub>9</sub>Co<sub>93</sub>O<sub>x</sub>, and Mn<sub>12</sub>Co<sub>88</sub>O<sub>x</sub> respectively, (D1-D4) SEM images of Fe<sub>3</sub>Co<sub>97</sub>O<sub>x</sub>, Fe<sub>6</sub>Co<sub>94</sub>O<sub>x</sub>, Fe<sub>9</sub>Co<sub>93</sub>O<sub>x</sub>, and Fe<sub>12</sub>Co<sub>88</sub>O<sub>x</sub> respectively.



Figure 2. The effect of Mn (Fe) doping and Au loading on catalyst performance.

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#### References

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