# The influence of modifiers ( $\mathbf{G a}$ and $\mathbf{Z r}$ ) on the performance of $\mathbf{C u Z n}$ catalysts for carbon dioxide hydrogenation to methanol 

Bongokuhle Xaba, ${ }^{\text {a,** }}$ Holger Friedrich, ${ }^{\text {a }}$ Abdul Mahomed, ${ }^{\text {a }}$ Sooboo Singh ${ }^{\text {a }}$<br>${ }^{\text {a }}$ University of KwaZulu-Natal, School of Chemistry and Physics, Durban, 4000, South Africa<br>*Corresponding author: 212509650@stu.ukzn.ac.za

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## 1. Introduction

Global warming is the conundrum of the twenty-first century and strategies that can convert $\mathrm{CO}_{2}$, the most prevalent greenhouse gas to valuable products such as methanol, a clean fuel are very attractive. ${ }^{1,2}$

## 2. Experimental (or Theoretical)

In this work zirconium and gallium promoted $\mathrm{Cu}-\mathrm{Zn}$ supported methanol synthesis catalysts were prepared by incipient wetness impregnation, characterized and evaluated in the conversion of $\mathrm{CO}_{2}$ to methanol. The catalytic testing was performed in the fixed-bed reactor.

## 3. Results and discussion



Figure 1. H2-TPR profiles


Table 1. The selectivities at isoconversion of $5.4 \%$ for $\mathrm{CuZnO}-\mathrm{ZrO}_{2} / \mathrm{Al}_{2} \mathrm{O}_{3}(\mathrm{CZZA})$ catalyst $\left(\mathrm{P}=20 \mathrm{bar}, \mathrm{T}=240^{\circ} \mathrm{C}\right.$ )

## 4. Conclusions

The catalytic results demonstrated that the Zr incorporated (CZZA) catalyst had the highest methanol productivity relative to the other two evaluated catalysts due it higher reducibility.

## References

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