Understanding 5-Hydroxymethylfurfural (HMF) Production Derived from Chitosan Using Solid Acids

Prasenjit Bhaumik, ^{1†} <u>Tai-Wei Tzeng</u>, ^{1‡} Guo-Chun Yang, ¹ Po-Wen Chung ^{1*}

E-mail: twtzeng72@gmail.com

Abstract

Selective chemical transformation of 5-hydroxymethylfurfural (HMF) from chitosan using various solid acids has been demonstrated in this study and we have discovered that the molar yield (%) of HMF chemically derived from commercial chitosan (chitosan-com) can reach 30±3% high while the reaction was carried out at 180°C for 2 h under cosolvent system. HMF production derived from solid acid A is comparable in comparison to the one from homogeneous catalysis systems using organic/inorganic acids shown in Figure 1. In addition, the tarry carbonaceous material was also produced upon the chitosan conversion which might impede the recyclability; however, solid acid A can be reused for chitosan conversion and it has shown 37% of recycling efficiency after regeneration.

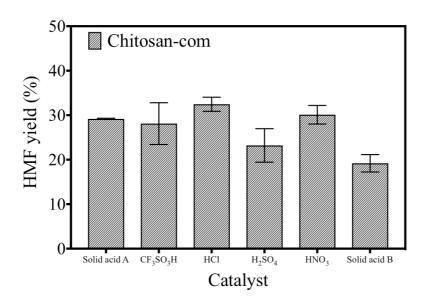


Figure. 1 The 5-hydroxymethylfurfural produced from commercial chitosan using heterogeneous and homogeneous catalysts.

¹ Institute of Chemistry, Academia Sinica, 128 Academia Road, Section 2, Nankang, Taipei 115, Taiwan.