# Novel low temperature $NO_x$ storage-reduction catalysts for diesel engine emissions

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**Abstract:** The promotional effect of Zr addition promoted the activity of Pt/BaO/CeO<sub>2</sub> catalyst for the NO<sub>x</sub> storage reduction was investigated. The Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> catalyst with the Ce/Zr molar ratio of 1:9 showed excellent NO<sub>x</sub> removal activity along with good durability. The superior activity of the Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> catalyst can be attributed to the greater amount of oxygen vacancies on the catalyst surface.

**Keywords:** NO<sub>x</sub> storage reduction, Ce-Zr mixed oxides, low temperature

#### 1. Introduction

The removal of  $NO_x$  emission from lean-burn engine exhaust has attracted much attention in environmental catalysis, and one of the most promising technologies is  $NO_x$  storage reduction  $(NSR)^1$ . NSR is an inherent transient operation in which the feed gases to the reactor are periodically switched between fuel-lean and fuel-rich gases<sup>1</sup>. The NSR catalyst previous studied by us is  $Pt/BaO/CeO_2$ , which can effectively remove  $NO_x$  in the  $200-400^{\circ}C$  temperature range, performing relatively poorly at lower temperature due to both limited trapping and regeneration efficiencies<sup>2</sup>.  $NO_x$  emissions during engine cold start make up a significant portion of total  $NO_x$  emissions during test cycles, with the catalyst remaining below  $200^{\circ}C$  for a significant amount of time<sup>3</sup>. Therefore, in order to improve overall emissions performance, one feasible approach is to incorporate components with good low-temperature  $NO_x$  removal ability into the  $Pt/BaO/CeO_2$  catalyst.

#### 2. Experimental

The CeZrO<sub>x</sub> nanomaterials were prepared by a hydrothermal method with different molar ratio of Ce to Zr (1:9, 3:7, 7:3). In addition, the CeO<sub>2</sub> and ZrO<sub>2</sub> were also prepared for comparison with the same method. Using the as-prepared oxide as supports, the 1wt.% Pt/8wt.% BaO/CeZrO<sub>x</sub> samples were synthesized by an impregnation method and then denoted as Pt/BaO/CeO<sub>2</sub>, Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub>, Pt/BaO/Ce<sub>0.7</sub>Zr<sub>0.3</sub>O<sub>2</sub>, and Pt/BaO/ZrO<sub>2</sub>, respectively. NSR cyclic measurements were conducted in alternating lean/rich atmospheres, while the lean feed contained 500 ppm NO and 8% O<sub>2</sub> and the rich feed contained 3% H<sub>2</sub>, both in balance N<sub>2</sub>. The lean phase and rich were fixed at 90s and 6s respectively, and the lean-rich cycles was 60.

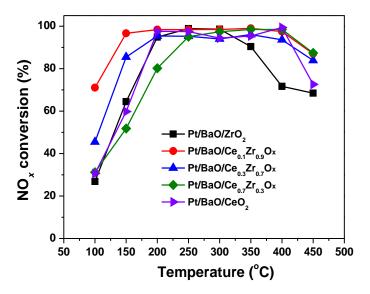
#### 3. Results and discussion

The durations of lean and rich phases of a dynamic NO<sub>x</sub> storage and reduction cycle are important operating parameters that not only affect the performance of the NSR catalyst, but also influence the fuel efficiency. In this study, the experimental cycle-averaged NO<sub>x</sub> conversion of NSR catalysts at different temperatures was summarized under alternating lean/rich condition in Figure 1. Apparently, the Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> catalyst, with a Ce/Zr molar ratio of 1:9, showed the best NO<sub>x</sub> removal efficiency. The NOx conversion over Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> at low temperature was greatly improved was high as 100% in a wide temperature range from 150 to 400°C. At the same time, the Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> went through 60 lean/rich cycles, and remained excellent performance. As shown in Table 1, Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> contained the largest amounts of oxygen vacancies. The superior activity of the Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> catalyst can be

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attributed to the more oxygen vacancies which was conducive to the adsorption and storage of  $NO_x$  in the lean phase, especially the storage at the low temperature.



**Figure 1.** NO<sub>x</sub> conversion of Pt/BaO/CeO<sub>2</sub>, Pt/BaO/CeO<sub>1</sub>, Pt/BaO/CeO<sub>2</sub>, Pt/B

**Table 1.** The Ce<sup>3+</sup> concentration of supports and catalysts.

Ce : Zr ratio	$CeZrO_x$	Pt/BaO/CeZrO <sub>x</sub>	Pt/BaO/CeZrO <sub>x</sub> -R <sup>[a]</sup>
$ZrO_2$			
$Ce_{0.1}Zr_{0.9}O_2$	20.3	19.5	20.1
$Ce_{0.3}Zr_{0.7}O_2$	18.9	18.2	18.9
$Ce_{0.7}Zr_{0.3}O_2$	15.5	15.7	15.3
$CeO_2$	13.5	13.6	14.0

 $\mathbf{R}^{[\mathbf{a}]}$ : catalysts were reduced by 3%  $\mathbf{H}_2$ .

#### 4. Conclusions

A novel NSR catalyst prepared by a hydrothermal method was used for  $NO_x$  storage reduction. The Pt/BaO/Ce<sub>0.1</sub>Zr<sub>0.9</sub>O<sub>2</sub> catalyst with a Ce:Zr molar ratio of 1:9 showed high  $NO_x$  removal efficiency and good durability. XPS results indicated that Zr addition increased the amount of oxygen vacancies, promoting  $NO_x$  storage at the low temperature.

### References

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